Course Syllabus

Please note: Normally, a syllabus states the rules of a course explicitly, but I wish to include additional lenience toward any of you who may be facing difficulties due to the various ongoing disruptions caused by the Covid-19 pandemic. Let's all work together with that thought in mind. We want to get it right. Let’s therefore recognize that some procedures related to your obligations to the course, including due dates, may need to be modified during the term.

Syllabus

General Biochemistry BB450/550 Winter 2021

Professor: Phil McFadden (he/him/his). I’m very OK with students calling me "Professor Phil" for short. But I will answer to just about anything.

My Email: phil.mcfadden@oregonstate.edu

Teaching Assistants (TA's):

Our two TA’s are experienced graduate students with biochemical research expertise!

• Jesse Howe howejess@oregonstate.edu
• Tilo Chatterjee chattert@oregonstate.edu

Undergraduate Learning Assistants (LA's):

Our LA’s took the course last year and must have wanted more!

• Saylor Miller millesay@oregonstate.edu
• Anjali Panikar panikarg@oregonstate.edu
• Mihir Palan palanm@oregonstate.edu
• Shrida Sharma sharmshr@oregonstate.edu

Office hours and contact information for Prof McFadden and our TA’s will be announced.

Course Prerequisites: Two terms of organic chemistry (CH 331 and CH 332 or CH 334, CH 335, and CH 336).

COURSE SCHEDULE and GRADING
Lectures by Prof McFadden will be held by Zoom, MWF at 9am to 9:50am Pacific Time. The Zoom link is given below as well as in the Zoom sidebar of your Canvas pages.

Zoom lecture link:

GENERAL BIOCHEMISTRY (BB_450_X001_W2021)
Every week on Mon, Wed, Fri, until Mar 12, 2021 (Except for MLK Day!)
https://oregonstate.zoom.us/j/99876839058?pwd=TEo5RDFka005OUdJVHFPVQVQ09
(Links to an external site.)
Zoom Password: 7s992e

Weekly recitations hosted by our TA’s and LA’s will provide help in understanding language, concepts and relevant problem solving. Zoom links to the recitations will be announced. Recitations begin in Week 1 and will be offered all ten weeks of the term.

The course will be built from week to week as follows:

Unit 1. Proteins

Includes Weeks 1, 2 and 3 which cover water, amino acids, protein structure and function


Unit 2. Enzymes

Includes Weeks 4, 5 and 6 which cover protein binding reactions, enzyme function, enzyme kinetics, enzyme mechanisms, and enzyme inhibitors

Main topics: Saturable oxygen binding by myoglobin and hemoglobin, including subunit-subunit cooperativity. Substrates, products and active sites of enzymes. Thermodynamics of enzyme reactions. Enzyme kinetics and inhibition. Example mechanisms of enzyme catalysis, including protease mechanisms.

Unit 3. Metabolism

Includes Weeks 7, 8, 9 and 10: Sugar metabolism, energy, and metabolic regulation

Main topics: Carbohydrate molecular structure (assumes background knowledge from prerequisite courses in organic chemistry). Carbohydrate fuel metabolism, including glycolysis, gluconeogenesis, and the glycogen fuel reserve. Energetics of oxidative pathways. Energy capture by ATP and electron carriers. Reciprocal strategies of
catabolism and anabolism. Signal cascades in carbohydrate metabolism. Enzyme regulation by covalent modification, particularly by protein kinases. Enzyme-mediated signal cascades.

**Final letter grades:** Canvas quizzes that are closely tied to the MWF lectures and weekly thought questions will be offered to you every week. Your performance on those ten quizzes will account for 75% of your course grade. Participation and attendance, homework and challenge questions from your weekly recitation will account for the other 25% of your course grade.

**LEARNING RESOURCES**

**Lecture notes:** Zoom lectures and the associated notes will be posted on Canvas. These notes will undoubtedly not include every detail of what I say, so I strongly suggest taking your own supplementary notes.

**Problem solving:** Thought questions (including verbal and numerical problems) will be posted on Canvas.

**Slack channels** for discussion will be available 24/7. Prof McFadden will visit his office hour channel on a regular basis. You can sign up and get to our Slack channels by accepting our invitation [here](https://example.com) (Links to an external site.).

**Recitation challenge problems** will be offered to you each week.

**Textbooks:** As references and source of figures and examples, we will refer to two free textbooks:

1. *Biochemistry Free for All* (2016) authored by Kevin Ahern, Indira Rajagopal and Taralyn Tan. You can find the latest formatting of the book [here](https://example.com) (Links to an external site.).


or if you prefer, you can borrow or purchase a copy of any edition of the Berg book, ranging between the 7th and the most recent (9th) edition.

**EXPECTATIONS AND COURSE LEARNING OBJECTIVES**
What are we looking for from you?

We expect you to diligently study vocabulary, concepts and problem-solving techniques related to the science of biochemistry. A complex topic like biochemistry requires considerable advanced reading and studying prior to attending lectures and recitations. A determined effort to listen actively and then taking sufficient time to study afterwards to put all the strands together is what is usually required to master the material.

What can you expect to learn? (please note that these lengthy objectives will be broken down into compact weekly learning objectives in the weekly Canvas modules)

- By the end of Course Unit 1 you can expect to be able to draw the covalent structures of all the amino acids as they exist in water as well as to draw how they are chemically linked together in proteins and affected by noncovalent bonds of several kinds. You should be able to recognize and describe the main primary, secondary, tertiary and quaternary structural elements in any protein molecule displayed in standard formats such as those in the biochemical literature and in publicly accessible internet sources such as the protein data bank (Links to an external site.). You should also be able to identify which of the functional chemical groups of protein molecules affect their folding and their denaturation (unfolding) in the watery conditions of living cells. As a case in point for quantitatively relating structure and function, you will learn how to use the Henderson Hasselbalch equation, which describes pH buffering, to predict protein charge and functional properties as the pH of a solution is varied. You can expect to be able to explain in conversational terms to anyone with minimal technicalities the structures of kinds of protein fibers that are familiar to all (e.g. collagen of tendons and bone, keratin of skin and hair) or perhaps not so familiar (membrane proteins, taste bud receptors, antibodies, myoglobin, actin filaments, microtubules). Given the “purification behavior” of a protein by chromatography and electrophoresis, you will be able to draw conclusions and state important aspects of the protein’s structure, including its mass as determined by mass spectrometry. You will also be able to sketch out the 3D structures of two globular proteins, myoglobin and hemoglobin.

- From Course Unit 2, you can expect to gain an appreciation of the machine-precision of protein function by conceptualizing both in language and in sketches how myoglobin and hemoglobin function as oxygen binding proteins. You will be able to explain how hemoglobin function is dramatically tuned by subtle structural changes, including minor amino acid substitutions, subunit-subunit cooperativity, and the binding of small molecules such as carbon dioxide. You will learn to name enzymes and describe their catalytic functions. You will be able to quantitatively portray any enzyme reaction as a chemical interconversion between substrates and products, with modulation by activators and inhibitors. You will use qualitative rules-of-thumb to compare the catalytic power of enzymes. You will also learn to use formulas and constant terms (Km and Vmax) in a steady state kinetic model (the Michaelis Menten equation) to quantitatively predict how the
speed of an enzyme reaction varies with the concentrations of substrates, products and inhibitors. By studying the mechanisms of a few well-understood enzymes, including those involved in cutting proteins, you will conceptualize which chemical features of an enzyme active site govern the making and breaking of bonds as substrate is turned to product.

- Course Unit 3 builds from the above concepts toward your understanding of how enzymes are functionally tied together into regulated metabolic pathways. Specifically, you will be able to sketch out pathways to describe how sugar molecules are metabolized by the two centrally important pathways, glycolysis and gluconeogenesis. The knowledge that you will be able to sketch in diagrammatic format will include the ten enzymatic reactions of glycolysis and the paths by which they work in concert in the cytoplasm of all organisms for the purpose of capturing chemical energy from food sugar. From there you will learn to question how various organisms and cells employ glycolysis to serve their varied needs for sugar fuel. By learning to appreciate the important distinction between catabolic and anabolic pathways, you can expect to be able to logically predict when (and why) various tissues such as liver and muscle turn-on and turn-off their specialized pathways of sugar metabolism (including glycolysis, gluconeogenesis, glycogen breakdown, and glycogen synthesis). By studying some examples of enzymes whose reaction is turned on and off according to physiological needs, you will be able to explain how allosteric regulation and regulation by post-translational covalent modification (specifically protein phosphorylation) affect the speed and function of enzymes. You will learn how cell receptors and intracellular proteins pass information to each other in signal transduction cascades, often involving protein kinase. You can expect to be able to trace the flow of several such cascades that are currently well understood. Finally, you will be able to account for the daily variations in our metabolism (between hunger and satiety, for example) by relating hormonal signals in the blood to intracellular signals and switches that operate at the level of protein phosphorylation reactions.

UNIVERSITY POLICIES

The goal of Oregon State University is to provide students with the knowledge, skill and wisdom they need to contribute to society. Our rules are formulated to guarantee each student's freedom to learn and to protect the fundamental rights of others. People must treat each other with dignity and respect in order for scholarship to thrive. Behaviors that are disruptive to teaching and learning will not be tolerated, and will be referred to the Student Conduct Program for disciplinary action. Behaviors that create a hostile, offensive or intimidating environment based on gender, race, ethnicity, color, religion, age, disability, marital status or sexual orientation will be referred to the Affirmative Action Office.

The Department of Biochemistry/Biophysics strictly follows the above university policy on student conduct. Behaviors disruptive to the learning environment will not be tolerated and will be referred to the Office of Student Conduct for disciplinary action.
Use of cell phones and excessive talking between neighbors is prohibited in the classroom.

**Student Conduct Regulations** are described at [http://studentlife.oregonstate.edu/code](http://studentlife.oregonstate.edu/code). Students are expected to be honest and ethical in their academic work. Cheating or plagiarism by students is subject to the disciplinary process outlined in the regulations. Dishonesty is defined as an intentional act of deception in one of the following areas:

- cheating- use or attempted use of unauthorized materials, information or study aids
- fabrication- falsification or invention of any information
- assisting- helping another commit an act of academic dishonesty
- tampering- altering or interfering with evaluation instruments and documents
- plagiarism- representing the words or ideas of another person as one’s own

**Accommodations for students with disabilities** are determined and approved by Disability Access Services (DAS). If you, as a student, believe you are eligible for accommodations but have not obtained approval please contact DAS immediately at 541-737-4098 or at [http://ds.oregonstate.edu](http://ds.oregonstate.edu). DAS notifies students and faculty members of approved academic accommodations and coordinates implementation of those accommodations. While not required, students and faculty members are encouraged to discuss details of the implementation of individual accommodations.

**Reach Out for Success**: University students encounter setbacks from time to time. If you encounter difficulties and need assistance, it’s important to reach out. Consider discussing the situation with an instructor or academic advisor. Learn about resources that assist with wellness and academic success at [oregonstate.edu/ReachOut](http://oregonstate.edu/ReachOut). If you are in immediate crisis, please contact the Crisis Text Line by texting OREGON to 741-741 or call the National Suicide Prevention Lifeline at 1-800-273-TALK (8255)

**OSU COVID POLICES**

To contribute to the health and safety of all OSU community members during the ongoing COVID-19 pandemic, and to align with federal and state regulations, executive orders, and guidance, the university has adopted two policies all community members are expected to observe:

- Policy on Face Coverings in Public and Common Settings ([https://policy.oregonstate.edu/UPSM/04-041_COVID19_face_covering](https://policy.oregonstate.edu/UPSM/04-041_COVID19_face_covering))
This policy requires faculty, staff, students and visitors across all OSU locations to use masks or cloth face coverings, or an appropriate alternative, when in enclosed OSU public and common areas, unless an exception is met; and

- Policy on Physical Distancing During Covid-19 Pandemic (https://policy.oregonstate.edu/UPSM/04-040_covid19_social_distancing (Links to an external site.))

This policy requires faculty, staff, students and visitors across all OSU locations to maintain six-feet of physical distance between others when in enclosed OSU public and common areas, unless an exception is met.

Please review and familiarize yourself with these policies and supplemental guidance (https://covid.oregonstate.edu/sites/covid.oregonstate.edu/files/face_covering_guidance_6-3-20.pdf (Links to an external site.)).

Thank you for contributing to the health and safety of the community. If you have questions about these policies, please contact me or you may submit further inquiries to the Coronavirus Question form (https://oregonstate.qualtrics.com/jfe/form/SV_cTpAHJzw4P3zyQd (Links to an external site.)).

COURSE POLICIES RELATED TO SYNCHRONOUS ZOOM ATTENDANCE

Zoom Classroom Norms

We invite everyone to participate in our scheduled synchronous lectures and recitations in Zoom. The recitations will hopefully be interactive with your participation; it is really not practical for our lectures (with up to 400 in attendance) to be as interactive.

Student-to-student interaction in recitations is highly valued and encouraged. To ensure that small group assignments in breakout rooms are productive and beneficial to everyone’s learning, please follow the lead of the TA’s and LA’s who are managing the recitations. Practice these widely accepted rules of social behavior:

- Listen actively to the viewpoints of group members; listen more than you talk, encourage equal participation.
- Practice reflective listening; restate and clarify what a group member is saying.
- Use academic and professional language to communicate your answers/contribution to group task.
- Have a positive attitude; provide constructive feedback to group members’ viewpoints.
- Weigh the feedback from peers; don’t reject it immediately.
- Do not be a passive observer; motivate, and help the group to reach an accurate consensus on the group task/product.
DIVERSITY AND INCLUSION STATEMENT

I have adopted and adapted Monica Linden’s diversity and inclusion statement (from Brown University). The sentiments she has expressed from the perspective of a research neuroscientist absolutely mirror my own personal perspective as a research biochemist. I have slightly altered her wording, with gratitude for her example of eloquence on this important aspiration.

In an ideal world, science would be objective. However, much of science is subjective and is historically built on a small subset of privileged voices. I acknowledge that the readings for this course were authored in the main by white men. Furthermore, the course often focuses on historically important biochemistry experiments which were mostly conducted by white men. Thankfully many have come to realize that integrating a diverse set of experiences is important for a more comprehensive understanding of science. Please contact me if you have any suggestions to improve the quality of the course. I would like to create a learning environment for students that supports a diversity of thoughts, perspectives and experiences, and honors your identities (including race, gender, class, sexuality, religion, ability, etc.) To help accomplish this:

- If you have a name and/or set of pronouns that differ from those that appear in your official OSU records, please let me know.
- If you feel like your performance in the class is being impacted by your experiences outside of class, please don’t hesitate to inform me. If you prefer to speak with someone outside of the course, Dean Henri Jansen, Associate Dean of the College of Science is an excellent resource who will maintain your anonymity.
- I (like many people) am still in the process of learning about diverse perspectives and identities. If something was said in class (by anyone) that made you feel uncomfortable, please talk to me about it. (Again, anonymous feedback is always an option).